

# Bi-Fe-Sb (Bismuth-Iron-Antimony)

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Recently, [2008Boa] determined an isothermal section at 250 °C and two vertical sections at 80 at.% Bi and 30 at.% Sb, respectively, for this ternary system. They also developed a thermodynamic description of the system.

## Binary Phases

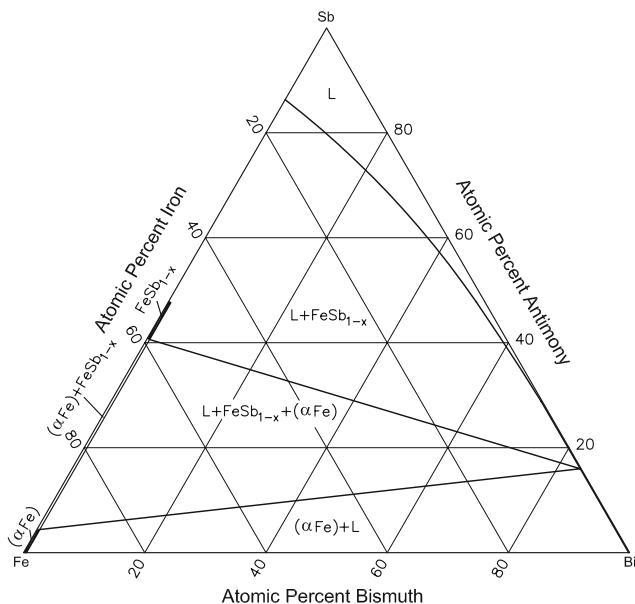
The Bi-Fe phase diagram depicts no intermediate phases. The mutual solubility between Bi and Fe is very limited. A continuous solid solution forms between the isostructural elements Bi and Sb. In the solid state, a miscibility gap occurs with the critical point at 225 °C and 63.7 at.% Sb [1992Feu]. In the Fe-Sb system, two intermediate phases are known:  $\text{FeSb}_{1-x}$  ( $B_8$ , NiAs-type hexagonal; also denoted  $\text{FeSb}$  or  $\epsilon$ ) and  $\text{FeSb}_2$  (orthorhombic). Using their own experimental results as additional input, [2008Boa] optimized the binary parameters for the Bi-Fe and Fe-Sb systems. The parameters for the Bi-Sb system were adopted from [1992Feu]. Computed phase diagrams for all the three binaries were presented by [2008Boa].

## Ternary Phase Equilibria

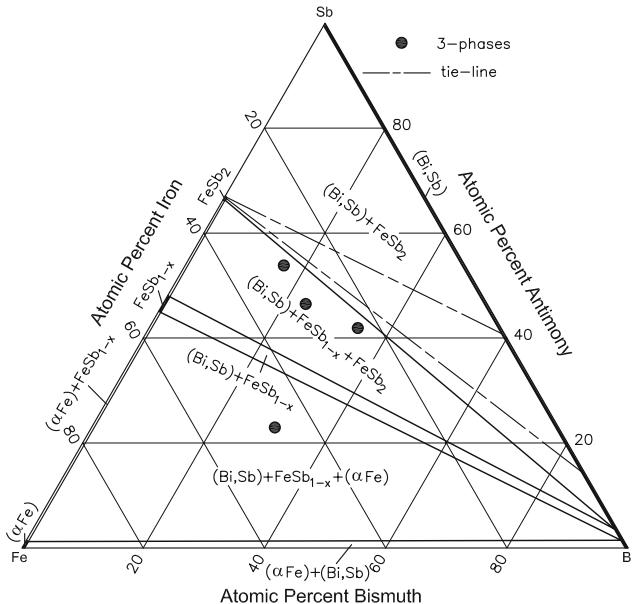
With starting metals of 99.999% Bi, 99.99% Fe, and 99.999% Sb, [2008Boa] melted in evacuated silica tubes 18

ternary compositions along the isoconcentrate lines of 80 at.% Bi and 30 at.% Sb. The final anneal for the isothermal study was at 250 °C for 6-9 months. To determine the thermal arrests, differential scanning calorimetry or differential thermal analysis was conducted at heating/cooling rate of 0.5-2 °C/min. The composition of the co-existing phases was determined by electron probe microanalysis.

Combining their new experimental results and those of [2006Boa] on the thermodynamic properties with the literature data, [2008Boa] optimized and listed the interaction parameters for this ternary system. Two isothermal sections at 792 and 250 °C and two vertical sections along 80 at.% Bi and 30 at.% Sb were computed. The isothermal section at 792 °C (Fig. 1) shows one three-phase field of  $(\alpha\text{Fe}) + \text{FeSb}_{1-x} + \text{L}$ .  $\text{FeSb}_2$  is not stable at this temperature. The section at 250 °C (Fig. 2) is compared with the experimental data obtained by the authors [2008Boa]. Figures 3 and 4 show the details of the vertical sections at 80 at.% Bi and 30 at.% Sb between 267 and 327 °C. Two invariant horizontals corresponding to the transition reactions  $U_1: \text{L} + \text{FeSb}_2 \leftrightarrow (\text{Bi}, \text{Sb}) + \text{FeSb}_{1-x}$  and  $U_2: \text{L} + \text{FeSb}_{1-x} \leftrightarrow (\text{Bi}, \text{Sb}) + (\alpha\text{Fe})$  are seen in Fig. 3 and 4. The experimental points [2008Boa] are shown. In addition, a ternary monotectic reaction M:  $\text{L}_1 \leftrightarrow \text{L}_2 + (\text{Fe}) + \text{FeSb}_{1-x}$  was computed by [2008Boa] at 991 °C.

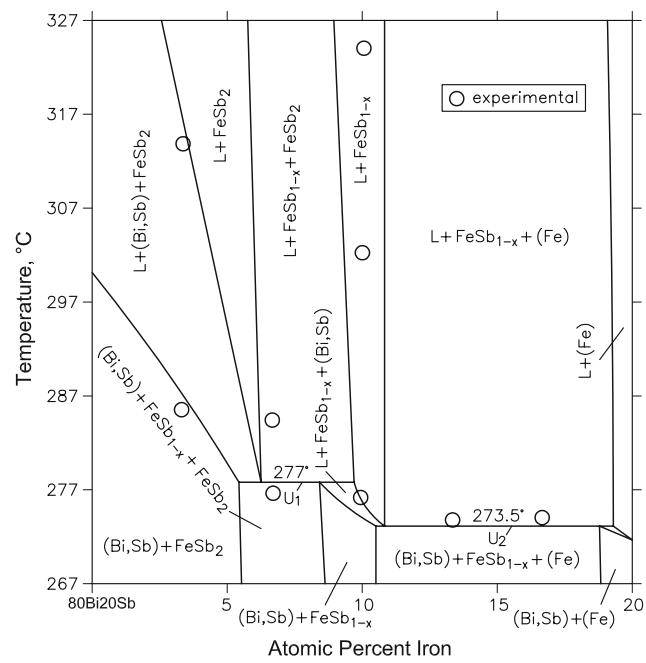


**Fig. 1** Bi-Fe-Sb computed isothermal section at 792 °C [2008Boa]

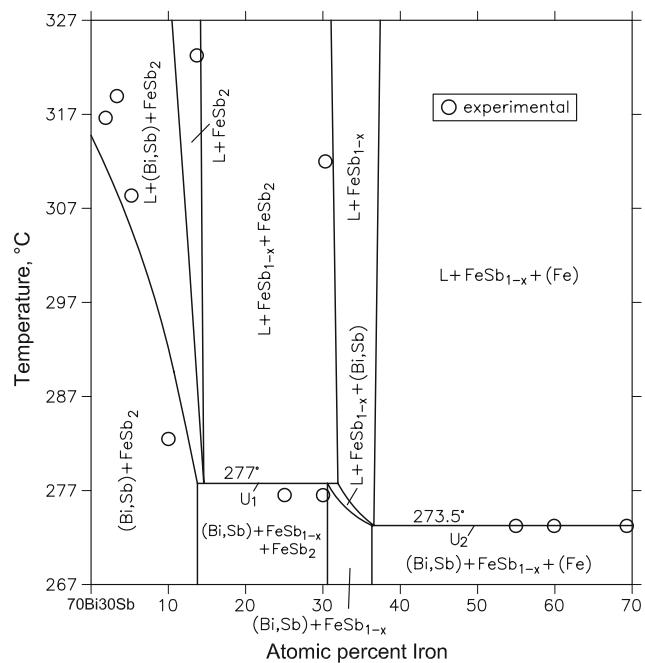


**Fig. 2** Bi-Fe-Sb computed isothermal section at 250 °C [2008Boa]

## Section II: Phase Diagram Evaluations



**Fig. 3** Bi-Fe-Sb computed vertical section at 80 at.% Bi [2008Boa]



**Fig. 4** Bi-Fe-Sb computed vertical section at 30 at.% Sb [2008Boa]

## References

- 1992Feu:** Y. Feutelais, G. Morgant, J.R. Didry, and J. Schnitter, Thermodynamic Evaluation of the System Bismuth-Antimony, *CALPHAD*, 1992, **16**(2), p 111-119  
**2006Boa:** D. Boa, S. Hassam, K.P. Kotchi, and J. Rogez, Thermodynamic Investigation of the Moderately Dilute

Liquid Bi-Fe-Sb Alloys, *Thermochim. Acta*, 2006, **444**(1), p 86-90

**2008Boa:** D. Boa, S. Hassam, G. Kra, K.P. Kotchi, and J. Rogez, The Ternary Bismuth-Iron-Antimony System: Experimental Phase Diagram Study and Thermodynamic Evaluation, *CALPHAD*, 2008, **32**, p 227-239